Demi-réaction de réduction	Potentiel (V)
F_2 (g) $+ 2\,e^-$ $\rightarrow$ $2\,F^-$ (aq)	$E^{\circ}_{Re} = +\,2{,}87$
Au^{3+} (aq) $+ 3\,e^-$ $\rightarrow$ Au (s)	$E^{\circ}_{Re} = +\,1{,}50$
Cl_2 (l) $+ 2\,e^-$ $\rightarrow$ $2Cl^-$ (aq)	$E^{\circ}_{Re} = +\,1{,}36$
Br_2 (l) $+ 2\,e^-$ $\rightarrow$ $2\,Br^-$ (aq)	$E^{\circ}_{Re} = +\,1{,}07$
Ag^+ (aq) $+ e^-$ $\rightarrow$ Ag (s)	$E^{\circ}_{Re} = +\,0{,}80$
Hg^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Hg (l)	$E^{\circ}_{Re} = +\,0{,}78$
Fe^{3+} (aq) $+ e^-$ $\rightarrow$ Fe^{2+} (aq)	$E^{\circ}_{Re} = +\,0{,}77$
I_2 (s) $+ 2\,e^-$ $\rightarrow$ $2\,I^-$ (aq)	$E^{\circ}_{Re} = +\,0{,}53$
Cu^+ (aq) $+ e^-$ $\rightarrow$ Cu (s)	$E^{\circ}_{Re} = +\,0{,}52$
Cu^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Cu (s)	$E^{\circ}_{Re} = +\,0{,}34$
$\mathbf{2\,H^+\ (aq)\ \ +\ 2\,e^-\ \ \rightarrow\ H_2\ (g)}$	$\mathbf{E^{\circ}_{Re} = +\,0{,}00}$
Fe^{3+} (aq) $+ 3\,e^-$ $\rightarrow$ Fe (s)	$E^{\circ}_{Re} = -\,0{,}04$
Pb^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Pb (s)	$E^{\circ}_{Re} = -\,0{,}13$
Sn^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Sn (s)	$E^{\circ}_{Re} = -\,0{,}14$
Ni^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Ni (s)	$E^{\circ}_{Re} = -\,0{,}26$
Co^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Co (s)	$E^{\circ}_{Re} = -\,0{,}28$
Fe^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Fe (s)	$E^{\circ}_{Re} = -\,0{,}44$
Cr^{3+} (aq) $+ 3\,e^-$ $\rightarrow$ Cr (s)	$E^{\circ}_{Re} = -\,0{,}74$
Zn^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Zn (s)	$E^{\circ}_{Re} = -\,0{,}76$
Cr^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Cr (s)	$E^{\circ}_{Re} = -\,0{,}91$
Mn^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Mn (s)	$E^{\circ}_{Re} = -\,1{,}18$
Al^{3+} (aq) $+ 3\,e^-$ $\rightarrow$ Al (s)	$E^{\circ}_{Re} = -\,1{,}66$
Be^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Be (s)	$E^{\circ}_{Re} = -\,1{,}85$
Mg^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Mg (s)	$E^{\circ}_{Re} = -\,2{,}37$
Na^+ (aq) $+ e^-$ $\rightarrow$ Na (s)	$E^{\circ}_{Re} = -\,2{,}71$
Ca^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Ca (s)	$E^{\circ}_{Re} = -\,2{,}87$
Sr^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Sr (s)	$E^{\circ}_{Re} = -\,2{,}89$
Ba^{2+} (aq) $+ 2\,e^-$ $\rightarrow$ Ba (s)	$E^{\circ}_{Re} = -\,2{,}91$